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S-14 1. Solubility of Ethanol in
Water Explain why ethanol ($\text{CH}_3\text{CH}_2\text{OH}$)

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is more soluble in water than is ethane (CH₃CH₃). Answer Ethanol is polar; ethane is not. The ethanol —OH group can hydrogen-bond with water. 2.

Calculation of pH from Hydrogen Ion Concentration
What is the pH of a solution that has an H concentration of
(a) 1.75×10^{-5} mol/L; (b) 6.50×10^{-10} mol/L; (c) 1.0×10^{-4} mol/L;

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