

Chemistry Paper 2 Essay May June 2014 Answers

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Chemistry Paper 2 Essay May

II. be stopped by thin paper. (ii) Give the charge on each of the radiations mentioned in 1(c) (i) above. (iii) What term is used to describe each of the following nuclear processes? I. Combination of two lighter nuclei to form a heavy nucleus II. Splitting of a heavy nucleus into two or more lighter nuclei III.

Chemistry Paper 2 (Essay) ,May/June 2008

(a) (i) What are acidic oxides? (ii) Give one example of each of the following oxides: I. acidic oxide; II. basic oxide; (III) amphoteric oxide; (III) neutral oxide. [5 marks] (b) (i) Define each of the following terms: (I) Heat of combustion; (II) Heat of neutralization. (ii) Write an equation to illustrate each of the terms in 7(b)(i) above. (iii) Given that the standard heat of combustion ...

Chemistry Paper 2 (Essay) ,May/June 2008

In (b)(ii), candidates correctly named the possible states in which water can exist as, solid, liquid and gaseous. In (c)(i), most candidates correctly stated Graham's law of diffusion however, some of them lost marks as they left the conditions under which the law holds while others wrote density instead of vapour density.

Chemistry Paper 2 (Essay) ,May/June 2008

The standard of the paper was high and compared favourably with those of previous years. The rubrics were clear, concise and unambiguous. The questions were well structured and within the scope of the examination syllabus. The marking scheme was detailed, clear and fair to all candidates.

Chemistry Paper 2 (Essay) ,May/June 2012

Question 2 (a) (i) Define saturated solution . (ii)The solubility of KNO₃ at 20°C was 3.00 mol dm⁻³ If 67.0g of KNO₃ was added to 250cm³ of water and stirred at 20°C, determine whether the solution formed was saturated or not at that temperature.

Chemistry Paper 2 (Essay) ,May/June 2010

You can find all CIE Chemistry IGCSE (0620 and 0971) Paper 2 past papers and mark schemes below. For 0971 Cambridge says 'this syllabus is graded from 9 to 1 but is otherwise the same as Cambridge IGCSE Chemistry - 0620.

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